Crystal structure, magnetic and thermal properties of the non-stoichiometric alloys $Tb_{0.95}Pd_{2.20}Mn_{0.85}$ and $Dy_{1.00}Pd_{2.15}Mn_{0.85}$

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Abstract

The alloys RNi_2Mn and RCo_2Mn with R representing the rare earth element are isostructural analogues of the binary Laves phases RNi_2 , RCo_2 . We have synthesized the isoelectronic Pd-based RPd_2Mn analogues with R = Gd, Tb, and Dy, for the first time. They crystallize in a cubic structure, however, not in the cubic C15 (*Fd-3m*) structure of RNi_2Mn , nor the Laves phase' structure. Instead, the crystal lattice is described by the *Im-3m* space group with Tb/Dy, Pd, and Mn atoms randomly distributed over the 2*a* Wyckoff position of the W-type structure (A2, disordered bcc). The magnetization data reveal a standard paramagnetic behavior, with effective magnetic moments strikingly close to those expected for Tb³⁺ and Dy³⁺ free ions, down to 130 K.

Key words: $RETM_2$ Mn, RE_{1-y} Pd_{2+x+y}Mn_{1-x} alloys, solid solutions, intermetallics, crystal structure, magnetization

1. Introduction

 RT_2 Laves phases, where R stands for the rareearth element and T is the transition element, have attracted considerable and repeated attention. Generally, these Laves phases crystallize in a cubic structure of MgCu₂-type (C15, Fd-3m) or hexagonal structures of MgZn₂-(C14, $P6_3/mmc$) or MgNi₂-type (C36, $P6_3/mmc$), based rather on the principle of the close packing of spherical atoms rather than directiondependent (covalent) chemical bonding [1]. The empirical rules governing the formation of a Laves phase require the ratio of the atomic radii of R and T atoms to be between 1.05 and 1.67 [2]. However, there are exceptions to the rule. Considering the subject of the present work, we highlight the nominal RPd_2 series. Recently, the actual stoichiometry of the majority of the members was determined to be $R_{10}Pd_{21}$, and their crystal structure was identified as monoclinic of $Sm_{10}Pd_{21}$ -type (C2/m) [3].

 RT_2 alloys have been of scientific interest due to their remarkable magnetic properties and application potential. RFe₂ alloys exhibit large magnetic anisotropy and magnetostriction/magnetostrain, making them an excellent choice for magnetomechanical actuators and sensors, see [4] and references therein. RAl_2 and RCo_2 alloys reveal considerable magnetocalorical effects at low temperatures [5, 6], representing suitable materials for hydrogen liquefaction. In RMn₂ alloys, the Mn-Mn interatomic distance (critical limit 2.67Å), tightly connected with R atomic radius, determines in which members the manganese ions carry a magnetic moment [7] and in which they do not (for $R \geq \text{Ho}$) [8], of course, having a strong impact on the magnetic properties of alloys also through R-Mn interactions. RNi_2 alloys have been investigated as potentially reversible hydrogen storage [9, 10]. A discussion on whether the Ni moment is (non-)zero in RNi₂ has continued for decades; finally, a non-zero moment was observed using microscopic methods [11, 12].

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	$\mathrm{GdPd}_{2}\mathrm{Mn}$	$\mathrm{TbPd}_{2}\mathrm{Mn}$	$\mathrm{DyPd_{2}Mn}$
Composition – initial	1.00: 2.00: 1.03	1.00: 2.00: 1.03	1.00: 2.00: 1.03
Composition – as-cast sample	inhomogeneous \rightarrow $\rightarrow 0.81(2): 2.09(1): 1.10(1)$	0.95(3): 2.17(1): 0.88(1)	$0.93(3)\!:\!2.14(2)\!:\!0.93(1)$
Composition – annealed sample	inhomogeneous	0.95(3): 2.20(1): 0.85(1)	1.00(3): 2.15(1): 0.85(1)
$T_{\rm melt}$ (K)	1126/1230(5)	1321(4)	1366(3)
a (Å)	_	3.374(2)	3.315(1)
$\Theta_{\rm CW}$ (K)	_	-40(2)	-51(2)
μ_{eff} (μ_{B})	_	9.63(3)	10.60(2)
μ_{9T}^{4K} (μ_{B})	_	2.25(2)	2.98(2)
$\mu_{9T}^{100K}(\mu_{B})$	_	1.42(2)	1.89(2)
$\gamma (\text{mJ mol}^{-1} \text{ K}^{-2})$	_	11.6(3)	18.5(3)
β (mJ mol ⁻¹ K ⁻⁴)	_	0.292(2)	0.284(2)
$\Theta_{\rm D}$ (K)	-	71(1)	69(1)

T a ble 1. Characterization of samples R_{1-y} Pd_{2+x+y}Mn_{1-x}: initial composition, stoichiometry of the as-cast and annealed ingots; melting temperature T_{melt} ; lattice parameter a of the cubic Im-3m structure (error based on the least-square refinement of peaks positions); paramagnetic Curie-Weiss temperature Θ_{CW} ; effective magnetic moment μ_{eff} ; the value of the magnetization in 9 T determined from $M(\mu_0 H)$ data measured at 4 and 100 K; the electronic specific heat coefficient γ , Debye coefficient β , and the Debye temperature Θ_{D} are presented

During the ongoing research of RT_2 alloys, their physical properties have been tuned by substitutions (hundreds of publications). Significant attention has been dedicated to the RNi_2Mn alloys with R = Tb, Dy, Ho, and Er, e.g., [13–15]. The alloys crystallize in a cubic MgCu₂-type structure (C15, Fd-3m) identical to the RNi₂ Laves phase. Mn atoms are randomly distributed at the R and Ni positions (Wyckoff positions 8a and 16d, respectively); however, the Mn occupation of sites was reported not uniform - instead, the 8a site contains several percent of vacancies [13, 14]. The alloys with light R were not synthesized because of the highly different atomic radii of R and Mn, resulting in phase competition. A change in crystal structure was reported for high concentrations of Mn, e.g., to rhombohedral PuNi₃-type structure (R-3m) for RNi_2Mn_x alloys with R = Gd (x > 0.4), Tb (x > 1.25), Dy (x > 1.5), and Er (x > 2) [15–18]. The manganese content considerably affects the magnetic properties of the alloy; noticeably, the magnetic ordering temperature is multiple times higher than the RNi_2 analogues [14, 18]. Simultaneously, the magnetic moment and magnetostriction of the alloys decrease, and the coercive field increases with Mn substitution. Contrary to the small moment of Ni [19], which makes the R-R magnetic interaction by and large dominant in RNi_2 , the Mn moment in RNi_2Mn is considerably larger (about 1.4 $\mu_{\rm B}$) [20, 21], and therefore the *R*-T and T-T interactions play a role. A noncollinear ferrimagnetic structure in RNi_2Mn (with R = Tb) was proposed based on magnetization and neutron diffraction studies [17].

The structural and magnetic properties of RNi_2Mn alloys have inspired further studies on RCo_2Mn counterparts. Similarly, a noticeable increase in ordering temperature and a decrease in the total magnetic moment with increasing Mn content have been reported [22, 23]. Moreover, by Mn doping, a tuneable thermal expansion from negative to positive was observed in this series [24]. Following these studies, we have synthesized and characterized, for the first time, the heavy rare earth RT_2 Mn alloys with T = Pd having a larger atomic radius (= 1.37 Å, compared to 1.24 Å for T = Ni) whilst being isoelectronic with Ni.

2. Sample preparation and experimental details

Polycrystalline RPd_2Mn (R = Gd, Tb, Dy) alloys were prepared by arc-melting of the initial elements (Mn - 3N5 = 99.95% purity (AlfaAesar), Pd - 4N5, Gd - 3N, Tb - 3N, Dy - 3N). 3% of Mn was added to the content [25, 26]; the initial stoichiometry of the samples is listed in Table 1. The surface of the initial manganese pieces was etched with a dilute solution of nitric acid to remove any oxide layer before the initial weighting. Ingots of mass ~ 1 g were remelted three times to ensure a good volume homogeneity of the samples. The mass loss of the samples was approximately 5%. Based on the results presented below, Mn was heavily evaporated during the melting process.

The crystal structure, homogeneity, distribution of individual elements, and phase purity of synthesized samples were investigated employing X-ray diffraction (XRD) and electron microscopy. Part of the sample was ground and examined using a Bruker D8 diffractometer with Cu K_{α} radiation (Bragg-Brentano geometry, zero-diffraction Si plate as a holder). Rietveld analysis of the measured diffraction patterns was performed using the FullProf package [27]. Another part of the sample was polished and inspected by the scan-



Fig. 1. Electron microscopy on RPd_2Mn . BSE images and EDX elemental maps are presented for (a) as-cast and (b) annealed R = Gd; (c, d) annealed R = Tb; and (e, f) annealed R = Dy samples, highlighting the increased homogeneity of samples after annealing.

ning electron microscope MIRA, Tescan, equipped with the backscattered electrons (BSE) detector and the energy-dispersive X-ray (EDX) analyser. The data were analysed using the Esprit software (corrections for atomic number, absorption, and fluorescence were done).

The melting temperature of the sample and any possible additional transitions in the high-temperatu-

re region were examined by differential scanning calorimetry (DSC) on a SETSYS Evolution 24 (SE-TARAM Instrumentation). Al₂O₃ crucibles with a lid were used as sample holders. The DSC data were measured on ~ 10 mg samples under He protective atmosphere at temperatures from 320 to 1420 K with the heating/cooling rate of 10 K min⁻¹. The melting temperature, T_{melt} , was determined as the onset of the

DSC peak corresponding to the first-order transition from solid to liquid phase (Table 1).

With knowledge of the melting temperature, the samples were annealed at temperature $T_{\rm anneal} = 1273 \,\mathrm{K}$, approximately 50 K below $T_{\rm melt}$. The ingots were wrapped in Ta foil and sealed under an Ar atmosphere in quartz glass tubes. The ingots were annealed for one week at constant temperature and subsequently quenched in cold water. No contamination of the Ta foil or quartz tubes was observed. The annealed samples were again investigated by XRD, BSE, and EDX.

Magnetization and specific heat were measured on a Physical Properties Measurement System (PPMS, Quantum Design) on ~ 200 and ~ 12 mg samples. The magnetization measurement was carried out using a vibrating sample magnetometer (VSM) insert at temperatures ranging from 1.8 to 400 K. Specific heat was measured using a standard time-relaxation method in the temperature range from 1.8 to 300 K.

3. Sample characterization

The quality of prepared samples, their purity and stoichiometry were investigated by electron microscopy and powder X-ray diffraction. BSE and EDX techniques revealed that the TbPd₂Mn and DyPd₂Mn are single-phase samples. However, their stoichiometry differs from the desired one R:Pd:Mn = 1:2:1; see Table 1. Pd-rich samples, slightly R- and considerably Mn-deficient samples, were synthesized. The stoichiometry deviation is also reflected by the samples' mass loss of approximately 5%. The attempt to synthesize the GdPd₂Mn alloy resulted in a highly inhomogeneous sample (Fig. 1). The results of the EDX analysis were confirmed by X-ray diffraction patterns (discussed below).

Subsequently, the prepared samples were measured by DSC. The inhomogeneity of the GdPd₂Mn sample is also reflected in the DSC data (Fig. 2); several high-temperature peaks are observed. Notably, the corresponding peaks are also followed in the DSC data measured on cooling (indicating phase/s stability). Relatively sharp, individual peaks related to the solid-liquid transition in TbPd₂Mn and $DyPd_2Mn$ data are observed. T_{melt} for individual compositions was determined and is listed in Table 1. Low-intensity peaks are traced at around 1050 and 1184 K in the DSC data for TbPd₂Mn and at 982 K for DyPd₂Mn. We ascribe these peaks to the secondary phases in the samples, which are, nevertheless, present in relatively low amounts – they were not identified by EDX and XRD techniques within the resolution/background of these techniques. However, the peaks are not observed in the cooling data, suggesting their thermal instability



Fig. 2. DSC measurement on as-cast *R*Pd₂Mn samples. The inset highlights minor intensity feature marked in the main panel ascribed to the impurity phase.

and, therefore, the importance of the sample annealing.



Fig. 3. Powder X-ray diffraction patterns of DyPd₂Mn. Fits the *Im-3m* model crystal structure (red line), difference curve (blue line), and Bragg positions (green marks) are presented.

Indeed, the annealing process slightly improved the quality/homogeneity of the TbPd₂Mn and DyPd₂Mn samples. The GdPd₂Mn sample was also changed by annealing; however, it remained highly inhomogeneous (Fig. 1b). This sample was excluded from further investigation. The stoichiometry of the R = Tb and Dy members was influenced by annealing. The manganese content in the samples further decreased, notably to the same value for both members (Table 1). The stoichiometry of the samples $R_{1-y}Pd_{2+x+y}Mn_{1-x}$ is shifted from the ideal RPd_2Mn composition, similar to the RPd_2 (or $R_{10}Pd_{21}$) case [3]. The atomic radii of the constituent elements have a considerable impact on the actual stoichiometry and, as demonstrated below, on the crystal lattice of the RT_2Mn alloys.

4. Crystal structure of $TbPd_2Mn$ and $DyPd_2Mn$

The powder XRD patterns of the prepared samples are not consistent with the cubic C15 crystal structure, which has been reported for the analogues RNi_2Mn and RCo_2Mn . TbPd₂Mn and DyPd₂Mn are not ternary Laves phases. However, they cannot also be described by the monoclinic Sm₁₀Pd₂₁-type of structure (C2/m) reported for RPd_2 [3]. The XRD pattern for the DyPd₂Mn sample is presented in Fig. 3, and the analysis leading to the identification of its crystal structure is provided:

(i) The Le Bail method was used to determine the lattice parameters and corresponding crystallographic system from the diffraction patterns. The five most intensive peaks from diffraction patterns were selected, and the Dicvol program [28] from the FullProf package [27] was employed for the structure analysis. The program suggested three sets of lattice parameters and crystallographic systems: $a_c = 3.316$ Å for the cubic system (holohedral space group Pm3m); $a_h = 2.707$ Å and $c_h = 3.315$ Å for the hexagonal system (holohedral space group P6/mmm); and $a_t = 3.317$ Å and $c_t =$ 2.707 Å for the tetragonal system (holohedral space group P4/mmm). The suggested models were used to fit the diffraction patterns, and the agreement between the data and fit was evaluated. All three models describe all unambiguously observable (including lowintensity) peaks in the diffraction patterns. Of course, zero intensity in diffraction patterns at numerous reflections indicated by the model is observed since the holohedral groups have no systematic extinctions.

(ii) The space group describing the patterns was searched for using the CheckGroup program [27]. We note that the number of observable peaks in our diffraction patterns is relatively low for precise analysis. Naturally, the analysis was commenced with the cubic system as the most symmetrical one. Five noncentered space groups were suggested by the program, including the most symmetrical Pm-3m group. Fitting the diffraction pattern with this space group still showed many reflections without corresponding peaks in the data. Similar results were seen whilst testing the other suggested space groups. A centering of the lattice is necessary to reduce the number of reflections. Therefore, we tested supergroups of Pm-3m with fcc and bcc centering. Multiple additional reflections without corresponding peaks in the data were observed for the Fm-3m model (considering the lattice parameter $2 \times a_{\rm c}$; DyPd₂Mn does not belong among the so-called Heusler alloys, as seen when Mn is replaced by a larger radius p-block element [29]. The body-centered Im-3m space group describes all the measured peaks, and importantly, no additional reflections (without corresponding peaks in the data) are generated.

(iii) Determining the atomic positions of individual elements in a given space group is generally a nontrivial task. Several structure models – based on the formula unit, elements, and an occupational number of individual Wyckoff positions – have been devised and compared with the measured data (peak intensities). We started with the simplest model, a complete disorder of Dy, Pd, and Mn atoms on the single Wyckoff position 2a: (0, 0, 0). Almost perfect agreement is observed between the model (a = 3.315(1) Å) and the data.

A final fit of the model to the data was performed. Trials to fit the occupancies of individual elements on the 2*a* position failed. Therefore, the content of elements was fixed to the values determined by EDX analysis. The final fit of the data is presented in Fig. 3, and the refined lattice parameter is listed in Table 1. TbPd₂Mn diffraction data were fitted to the same model, leading to a similar agreement ($\chi^2 = 2.2$, com-



Fig. 4. Magnetization measurements on TbPd₂Mn and DyPd₂Mn alloys. The temperature dependence of magnetization M/H(T) and its reciprocal value H/M(T) are presented together with the fit to the Curie-Weiss law (within the 130–400 K interval; blue line) and its extrapolation to the low temperature (dashed line). Selected isothermal magnetization data are presented in the insets.

parable to $\chi^2 = 2.0$). We note that the diffraction patterns contain a relatively small number of peaks and high background, limiting our crystal structure determination. The uniform filling of a single site by three atoms with significantly different atomic numbers (atomic radii, atomic mass) is unusual. Nevertheless, based on the diffraction patterns, we have ruled out the crystal structures previously reported for related alloys (see section Introduction).

The complete atomic disorder on a single Wyckoff position in the *Im-3m* lattice makes it extremely difficult to understand and model the properties of the *R*Pd₂Mn alloys quantitatively, and thus effective structure-property relations are challenging to obtain.

5. Magnetic and thermal properties

The physical properties of the samples were in-

vestigated by magnetization and specific heat measurements. The temperature dependence of magnetization reveals a standard paramagnetic response of TbPd₂Mn and DyPd₂Mn at a temperature from 400 K to approximately 130 K (or even down to approximately 70 K, considering the uncertainty of the measured data, Fig. 4). No clear indications of ferrimagnetic ordering (R-Mn coupling), reported for RNi_2Mn and RCo_2Mn analogues [14, 18], are observed in this temperature interval. The attempts to describe the inverse dc-magnetic susceptibility by the hyperbolic form of the Curie-Weiss law for two sublattices [30–32] failed due to the slight curvature of the H/M curves. The H/M(T) dependence is described by a standard Curie-Weiss law with fitted paramagnetic Curie-Weiss temperatures, Θ_{CW} , and effective magnetic moment, μ_{eff} , listed in Table 1. The fitted values of μ_{eff} are strikingly close to those of the free ions of Tb^{3+} (9.72 μ_B) and Dy^{3+} (10.65 $\mu_{\rm B}$), indicating the paramagnetic state of the alloys and no significant Mn moment. The negative Curie-Weiss temperature suggests average antiferromagnetic exchange interactions. A bifurcation of zero field-cooled (ZFC) and field-cooled (FC) magnetization below approximately 70 K, presented in Fig. 5, is not typical for a purely antiferromagnetic material. We note that the temperature at which the ZFC and FC magnetization curves initially bifurcate is similar in TbPd₂Mn and DyPd₂Mn. One can speculate that it is connected with the magnetic interactions between Mn moments and, in turn, with the *R*-Mn coupling. However, the Mn moments should also be manifested in higher-temperature magnetization data. That is, values of μ_{eff} should differ from (be higher than) the values of R^{3+} free ions. We cannot rule out a possible presence of secondary phase/s, namely RMn_2 , although they were not observed by electron microscopy and X-ray diffraction. Both TbMn₂ and DyMn₂ reveal magnetic phase transition below about 40 K [33, 34], which is relatively close (about 20 K below) to the temperature of bifurcation of ZFC and FC magnetization curves. On the other hand, a second magnetic transition below 25 K in DyMn_2 [34] is not pronounced in the measured data.

In contrast, the TbPd₂Mn magnetization data reveal a second, relatively sharp anomaly at around 10 K, which is not observed in the DyPd₂Mn data. The anomaly might be connected with magnetic correlations between Tb-Tb, Tb-Mn, or Pd-Mn. A long-range magnetic ordering (antiferromagnetic type) is not expected, considering a complete disorder of constituent atoms on a single Wyckoff position. Alternatively, the presence of secondary phase/s, namely Pd-Mn alloys exhibiting ferromagnetic ordering at around 10 K [35], can be speculated about (not confirmed by the electron microscopy and X-ray diffraction measurements).



Fig. 5. Magnetization data measured under the ZFC and FC regimes in a magnetic field of 0.01 T. The insets contain the temperature evolution of the specific heat measured on the respective alloys.

Based on measured data, no unambiguous conclusion can be made.

The scenario of secondary phase(s) in the samples is supported by the specific heat measurements. A $C_{\rm p}/T$ vs T^2 plot is presented in the inset of Fig. 5. Only subtle anomalies are observed below 70 K. That is, any magnetic ordering of the RPd_2Mn alloys is not reproduced in the data. The low-temperature specific heat data (2–15 K) were analysed using the Sommerfield-Debye formula: $C_{\rm p}(T) = \gamma T + \beta T^3$ with γ standing for the Sommerfield coefficient of electronic specific heat, and β is Debye lattice (electron– phonon) specific heat coefficient. The fitted parameters, including the Debye temperature calculated as $C_{\rm p} = (12R\pi^4)/5(T/\Theta_{\rm D})^3$ [36], where R is gas constant, are similar for both alloys (listed in Table 1). Also, these parameters suggest that the two alloys do not magnetically order down to low temperatures. The anomalies in magnetization data (and weak anomalies in specific heat data) can be ascribed to secondary phase/s.

The possibility of a ferromagnetic ground state is mostly excluded in both alloys. The isothermal magnetization data (insets of Fig. 4) reveal hysteresis with a coercive field of about $0.24 \,\mathrm{T}$ for $\mathrm{TbPd_2Mn}$ and 0.36 T for DyPd₂Mn at 4 K. However, the magnetization does not saturate in magnetic fields as high as 9 T. The magnetization value in 9 T is significantly lower (Table 1) than the spontaneous magnetic moment expected for rare earth ions (9 $\mu_{\rm B}$ for Tb³⁺ and $10 \,\mu_{\rm B}$ for Dy³⁺). Hence, ferrimagnetic ordering can be speculated about. Nevertheless, the apparent complete atomic disorder (on 2a Wyckoff site), significant off-stoichiometry, and the probable presence of secondary phase/s strongly affect the low-temperature $(< 70 \,\mathrm{K})$ magnetic properties of the investigated samples, making their interpretation ambiguous.

6. Conclusions

Non-stoichiometric, considerably Mn-deficient $(x = 0.15), R_{1-y} Pd_{2+x+y} Mn_{1-x}$ samples with R =Tb and Dy were synthesized and characterized. The synthesis of the R = Gd analogue failed, demonstrating the difficulties in preparing new RT_2 Mn alloys; that is, the significance of the atomic radii ratio of constituent elements. Unlike the analogous alloys RNi_2Mn and RCo_2Mn crystallizing in a C15 type of structure (Fd-3m), the crystal structure of the investigated samples was determined to be fully disordered: R, Pd, and Mn atoms randomly distributed over the 2a Wyckoff position of the W-type structure (A2, bcc, Im-3m). The samples remained paramagnetic down to at least 130 K. Negative Curie-Weiss paramagnetic temperature and effective magnetic moment noticeably close to the value expected for free rare-earth ions were refined. No contribution of Mn magnetism was traced. A bifurcation between ZFC and FC magnetization below 70 K could be ascribed to magnetic correlations between R or/and Mn moments, particularly in the case of R = Tb where additional ZFC--FC splitting is observed below an additional anomaly at 10 K. Possible ferrimagnetic ordering/interactions were supported by isothermal magnetization data. However, the confident explanation of magnetization data remains ambiguous considering the atomic disorder (frozen metastable phases are investigated, 3rd law of thermodynamics) and non-stoichiometry of the investigated samples. Moreover, the possible presence of secondary phase/s, although they were not directly observed investigating the samples employing electron microscopy and X-ray diffraction methods, could also explain the observed bifurcation.

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